



Metabolism of Minerals

Shuzan Ali Mohammed

Medical Biochemistry & Molecular Biology

Faculty of Medicine - Benha University

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Magnesium (Mg): Sources: 1. Animal: Liver, kidney, heart, meat, & fish. 2. Plant: Vegetables especially chlorophyll, legumes & nuts.

Absorption:

- * About 40% of the ingested Mg is absorbed
- * usually from the **upper** small intestine.
- * Absorption of Mg is affecting by the factors affecting Ca absorption

Distribution:

- Mg is distributed in skeleton, tissues, & body fluids
- The total body Mg is about **21 gm**.
- About 70% in bone,
- 30% in tissues & body fluids (mainly intracellular)
- Mg concentrat. in muscles: 110 times that of plasma

Blood magnesium:

- The plasma Mg level is 2.2 ± 0.5 mg/dl.
- It is diffusible (80%) & non- diffusible form (20%).
- Its concentration in RBCs is higher than plasma.

Factors affecting plasma magnesium:

1. Aldosterone:

- Aldosterone \rightarrow Na & H₂O retention \neg
- Aldosterone \rightarrow K & Mg excretion

Hypomagnesaemia

2. Parathyroid hormone (Parathormone):

- Parathormone $\rightarrow \uparrow$ Mg absorption from intestine.
- Parathormone → ↑ Mg²⁺ mobilization from bone;
 (The net effect is increased plasma Mg level.)
- Prolonged hyperparathyroidism → depletes Mg stores
 Hyperparathyroidism → hypomagnesaemia and tetany
- **3. Kidney function:** Plasma **Mg increases** in **renal failure** due to failure of its excretion.

N.B.: In renal failure, the followings occur;
Hypocalcaemia, why? Due to loss of activation of vitamin D at C1 by 1 α-hydroxylase
Hyperphosphataemia (failure of excretion)
Hypermagnesaemia (failure of excretion)

Functions:

- **1. Mg enters in the structure of bone.**
- 2. Mg is important for normal contraction of muscle. Hypomagnesaemia → muscle weakness & paralysis
- 3. Mg is important for transmission of nerve impulse. Hypomagnesaemia \rightarrow somnolence & anesthesia
- 4. Mg decreases neuromuscular excitability; so, Hypomagnesaemia → tetany, (cannot be treated by calcium injection)
- 5. It activates many enzymes as: (phosphorylases, kinases, & phosphatase)

Excretion:

- In feces (70%): Most of unabsorbed Mg is excreted in the form of Mg phosphates.
- In urine (30%): About 150 mg/day of Mg is excreted through urine.

Requirements:

- RDA for adult is 0.3 gm/day.
- Dietary deficiency is rare in man.

Sodium (Na), potassium (K), & chloride (Cl): Sources:

- Salt of diet (Table salt, NaCl) is present in cheese, milk bread, cereals, legumes, nuts & some vegetables.
- Potassium is present in vegetable fruits, potatoes, liver, kidney, heart, meat, & fish.

Absorption:

- Absorption of Na mainly in the ileum.
- Absorption of K mainly in upper small intestine

Sodium (Na), potassium (K), & chloride (Cl): Distribution:

- ~ 2/3 of total Na, K, & Cl are in tissues & body fluids
- The remaining one third is in the skeleton.

Blood level:

- Na⁺ and Cl⁻ are chiefly extracellular,
- K⁺ is chiefly intracellular ion.
- Plasma Na is 135 150 mmol/L.
- Plasma K is 2.5 5.0 mmol/L.
- Plasma Cl is 90 110 mmol/L.

1. Vomiting:

- Short period vomiting → hypochloraemia & alkalosis
- Prolonged vomiting → hyponatraemia, hypokalaemia and acidosis.

2. Diarrhea:

→ hyponatraemia, hypokalaemia & acidosis.

3. Fluid infusion:

- fluids rich in Na, K, & Cl → hypernatraemia,
 hyperkalaemia & hyperchloraemia.
- fluids poor in Na, K & Cl → hyponatraemia, hypokalaemia & hypochloraemia.

4. Sweating:

- Excessive sweating → water loss in proportion to Na & Cl → hypernatraemia & hyperchloraemia.
- If fluids are given without replacing NaCl, we get hyponatraemia, and hypochloraemia.

5. Diabetes insipidus:

 This type of diabetes has the same effect like sweating on Na & Cl levels.

6. Renal failure:

- Either acute or chronic → hyponatraemia,
- Hyperkalaemia is more in acute renal failure.
- Little changes of plasma K in chronic renal failure.

7. Diuretics:

Some drugs (e.g. diamox and chlorothiazides) → hyponatraemia, and hypokalaemia.

- 8. Suprarenal function:
- Hyperfunction (Cushing's syndrome & hyperaldosteronism): →
 - Hypernatremia (Na & H₂O retention) $\rightarrow \uparrow$ Bl P., &
 - K & Cl excretion, (hypokalaemia & hypochloraemia with subsequent alkalosis).

 Hypofunction (Addison's disease): → the opposite occurs (hyponatraemia with ↓ Bl P., hyperkalaemia & hyperchloraemia with subsequent acidosis).

- 9. Acid base balance:
- **a. Acidosis:** → hyperkalemia & hyponatremia
 The urine becomes alkaline.
 - This condition is called **paradoxical acidosis**.

a. Alkalosis: → hypokalemia & hypernatremia.
 The urine becomes acidic.
 This condition is called paradoxical alkalosis.

Sodium (Na), potassium (K), & chloride (Cl): Functions:

- Functions of Na & K:
- 1. Na⁺ is the main **extracellular cation**.
- 2. K⁺ is the main **intracellular cation**.
- 3. Na⁺ & K⁺ enter in the formation of **buffers**.
- 4. Na⁺ & K⁺ maintain Bl volume, osmotic pressure & Bl P.
- 5. Na⁺ & K⁺ are essential for **muscle contraction** & transmission of **never impulse**.

Functions of Cl:

- 1. Cl⁻ enters in the formation of **gastric HCl**
- 2. Cl⁻ activates amylase enzyme.
- 3. It helps in buffering of **carbonic anhydrase** in the plasma through **chloride shift phenomenon**.

Sodium (Na), potassium (K), & chloride (Cl): Excretions:

- Excretion of Na⁺, K⁺, & Cl⁻ is controlled by Aldosterone.
- The major route of excretion is urine (95 %)
- Feces and sweats are minor routes (only 5%).

Requirements:

- The RDA of K⁺ is 3 5 gm in the form of KCL
- The RDA of Na⁺ is 10 15 gm in the form of NaCl

Sources:

- 1. Amino acids: cysteine, cystine, & methionine.
- **2. Vitamins:** thiamine, lipoic acid, & biotin.
- 3. Sulfolipids.
- 4. Mucopolysaccharides.

Absorption:

- It occurs mainly from the small intestine.
- Inorganic sulfate is poorly absorbed

Blood level:

- Sulfur level: 2.5 mg/dl.
- RBCs contains glutathione, & ergothionine, in addition to sulfur containing compounds present in plasma.

Sulfur (S) Sulfur containing compounds in the body:



(N.B.): Ergothionine is present in liver, semen & erythrocytes.



(N.B.): Taurine is formed from partial oxidation of cysteine

Metabolism of sulfur-containing amino acids:

2. Conversion to other neutral sulfur compounds:

- Proteins (mostly keratin).
- Glutathione (formed of glycine, cysteine, & glutamic).
- Amino ethyl mercaptan (in COASH).
- Homocysteine (their sulfur being oxidized into inorganic sulfate).
- Thiocyanate (detoxification).
- Urochrome (present in urine as a peptide substance).
- The later 3 compounds are excreted in urine.

3. Excretion in urine:

Under normal conditions, the three sulfur containing amino acids are excreted in urine in little amounts.



1. Inorganic sulfate

They are mostly used in the biosynthesis of PAPS (3'-phosphoadenosine-5'-Phosphosulfate)



Adenosine-5'-pohosphosulfate + ATP -----> 3'-Phosphoadenosine-5'-phosphosulfate+ ADP



Excretion:



Excretion:

1- Unoxidized sulfur (10%):

- 3 sulfur containing aa (cysteine, cystine & methionine)
- 3 sulfur containing vitamins (lipoic acid, biotin & thiamine)
- 3 sulfur compounds (urochrome, thiocyanate & mercupturic acid).
- 2- Oxidized sulfur (90.0%):
- Inorganic sulfates (80%): e.g. Na & K sulfate.
- Ethereal sulfates (10%): detoxification of phenols & alcohols, e.g. indican (indoxyl K sulfate), skatoxyl K sulfate, phenol sulfate & steroid hormone sulfate.

Requirements: Diet adequate in protein is sufficient.

Good Luck